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Law, Boyle's Law,

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Forces - Hydrogen  
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Dipole, Ion-Dipole,  
London Dispersion  
Interactions All  
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Worksheet # 1:

Introduction to Gas  
Laws. Earlier in  
your science  
education you  
learned to describe  
the gas state as the  
state of matter with  
no definite shape,  
no definite volume,  
and highly  
compressible. Now

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you have added the kinetic molecular theory to your understanding of gas behavior.

Accelerated  
Chemistry

b) Gas pressure = atmospheric pressure +  $h \cdot AP$  is a registered trademark of the College Board,

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1. Convert the following temperatures as indicated.
- a)  $0^{\circ}\text{C}$  to  $\text{K}$  \_\_\_\_\_  $273 \text{ K}$  \_\_\_\_\_ e)  $1 \text{ atm}$  to  $\text{kPa}$   $101.3 \text{ kPa}$  (s.f. = 100) \_\_\_\_\_
- b)  $-10^{\circ}$  ...

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relationships between gas physical properties are described as gas laws. Some of these properties are pressure, volume, and temperature. These laws show how a change in one of these properties affects the others.

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Period Date THE  
IDEAL GAS LAW n  
PV = nRT where  
pressure in  
atmosphere volume  
in liters = number  
of moles of gas  
Universal Gas  
Constant = 0.0821  
atm/mol.K Kelvin  
tem erature 1. 2. U  
150 5. 7. 8. 9. How  
m will occupy a  
volume of 2.50

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liters at 1.20 atm  
and 25.0 °C? moles  
of O<sub>2</sub> = 0.001

## Answer Key

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Key - Mr. Smith's  
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volume in liters =  
number of moles of  
gas Universal Gas  
Constant = 0.0821



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atm/mol.K Kelvin  
temperature 1. 2. U  
150 5. 7. 8. 9. How  
m will occupy a  
volume of 2.50  
liters at 1.20 atm  
and 25 0 C? moles  
of o 0.001 Gas  
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MOLE PROBLEMS-

MOLAR VOLUME

OF A GAS We Have

Learned That A

Mole Is A Mass Of

Material And

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Answer Key

Number Of  
Particles. A Mole  
Can Also Be A  
Measure Of Volume  
When We Are  
Talking About  
Gases. You May  
Remember From  
Previous Science  
Classes That All  
Gases Behave  
Basically The Same  
As Far As The  
Physical ...

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CHEMISTRY ...

Base your answers  
to questions 33 and  
34 on the  
information below.

Scientists are  
investigating the  
production of  
energy using

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Chemistry 2 nuclei  
(deuterons) and  
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hydrogen-3 nuclei  
Answer Key  
(tritons). The

balanced equation  
below represents  
one nuclear  
reaction between  
two deuterons.  $2\ ^1_1\text{H} + 2\ ^1_1\text{H} \rightarrow\ ^3_2\text{He} +\ ^1_0\text{n} + 5.23 \times 10^{-13}\text{ J}$

Identify the type of  
nuclear reaction



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represented by the  
equation.

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B Moles Packet 159  
89 203 220. system  
that consolidated  
the capabilities of

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### ANSWER KEY

Boyle's, Charles'  
and Gay-Lussac's  
Gas Problems I. If a  
gas at occupies  
2.60 liters at a  
pressure of 1.00  
atm, what will be its  
volume at a  
pressure of 3.50

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atm? 2. A gas occupies 900.0 ml, at a temperature of 27.0 0 C. What is the volume at 132.0 0 C? 3. What change in volume results if 60.0 mL of gas is cooled from 33.0 oc to 5.00 ...

Guilford County  
Schools /

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Homepage  
AP Chemistry Test  
Gases Packet  
Answer Key

do not write on this test. Please use the answer sheet.

Multiple Choice 1)  
A sample of argon gas is sealed in a container. The volume is doubled, while the temperature is held constant. What has

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happened to the  
pressure on the  
gas? A) It is halved.  
B) It is reduced by  
1/4 th. C) It is  
doubled.

AP Chemistry Test  
(Chapter 5) Please  
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Gases Intro -  
PhET: Free online  
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Gas Laws

Worksheet atm =

760.0 mm Hg =

101.3 kPa = 760 .0

torr Boyle ' s Law

Problems: 1. If 22.5

L of nitrogen at 748

mm Hg are

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Compressed to 725 mm Hg at constant temperature. What is the new volume?

2. A gas with a volume of 4.0L at a pressure of 205kPa is allowed to expand to a volume of 12.0L.



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60417ccaf54c43

Answer Key